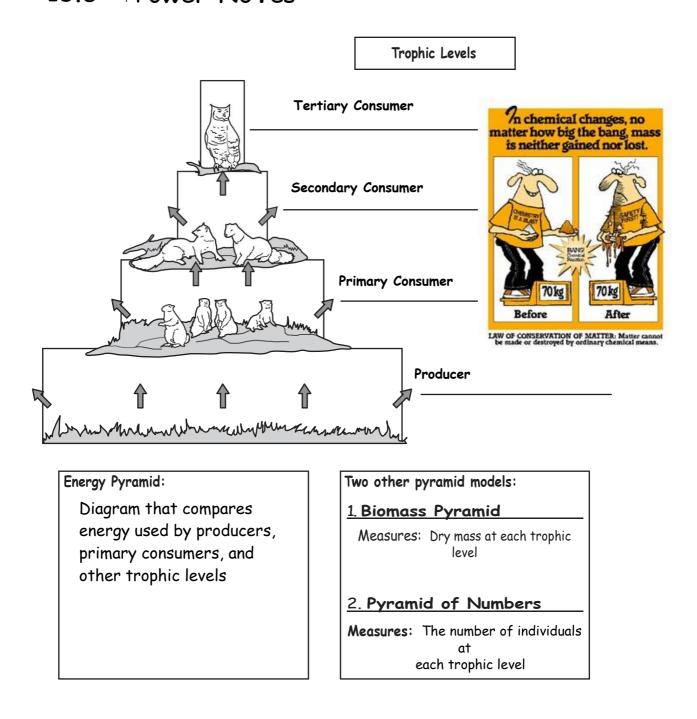
SECTION | PYRAMID MODELS 13.6 | Power Notes



The **law of Conservation of Energy** states that the total <u>energy</u> of an <u>isolated system</u> cannot change—it is said to be <u>conserved</u> over time. Energy can be neither created nor destroyed, but can change <u>form</u>, for instance <u>chemical energy</u> can be <u>converted</u> to <u>kinetic energy</u> in the explosion of a stick of <u>dynamite</u>.

The Law of the Conservation of mass: a fundamental principle of classical physics that matter cannot be created or destroyed in an isolated system, but can change <u>form</u>.

90% of the amount of available energy from one trophic level to another is lost as heat leaving only 10% of usable energy for organisms at each additional trophic level.